# The effect of ball milling on crystallite size 

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#### Abstract

Homogeneous mixing of Al , varying amounts of $\mathrm{Cu}, \mathrm{Mg}$ and Zn metal powders and SiC ceramic powders and mechanical alloys of metal powders by using high energy ball milling were carried out in the Retsch MM400 model mixer device, which performs movement in a spex manner. After this process, X-ray diffraction (XRD) was applied to the powdered mixtures. With the data obtained from XRD graphics; The crystallite size was calculated using the Scherrer equation, and the lattice stresses were calculated using the Williamson-Hall equation and comparisons between these two data were made. It was observed that the amount of Cu by weight, both the crystallite size, did not make a notable change for this property. Then, powder mixtures were sintered in hot isostatic press in argon atmosphere, which is a shielding gas, and turned into samples. These samples were polished and scanning electron microscopy (SEM) images were taken.


Key words: Powder metallurgy, ball milling, mechanical alloying, characterization, metal matrix composite

## 1. Introduction

Today, the need for light and low-cost materials in many industry applications tends to increase steadily. In line with this trend, new and more modern techniques are tried to be created for the production of metal matrix composites (MMK). These techniques include mechanical alloying (MA), which plays an active role in bonding metal-metal or metal-ceramic materials together. MA is a solid-state powder processing technique that allows powder particles to break, weld and then break again. The MA process is carried out using ball milling [1]. It has been used in the synthesis and balancing of new intermediate phases from 1985 to 2001 and in new processes that have not been tried so far until today [2]. John Benjamin and his friends described the MA process, considered as part of powder metallurgy, as a method for producing composite metal powders with controlled fine microstructure [3]. With the MA technique, the crystallite size of the matrix powder can be reduced to nano levels. Aluminum matrix composites find wide adaptation areas due to their high wear resistance, high elastic modulus, high hardness and high tensile strength aspects [4]. Ceramic reinforced aluminum matrix composites are frequently used in industries such as automotive and aerospace. These composites are used in piston rods, brake discs and cylinder blocks in automotive applications, and wing covers in aviation applications [5].

In the study of S. Kamrani et al., the average particle size -325 mesh Mg and 50 nm SiC powders were first mixed in the Turbula T2C device for 20 minutes in a $2 \mathrm{wt} \%$ stearic acid environment. Then, in Pulverisette 5, a planetary ball mill, these powders were alloyed at 250 rpm with zirconia
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balls in a polyethylene bottle for $1,3,5,15,20$ and 25 hours. In this process, $10: 1$ was used as the ball to powder weight ratio. They carried out all operations in an argon shielding gas atmosphere. They obtained the crystallite size in the Mg matrix from XRD peak broadening. As a result of this study, they showed that crystallite size decreased with increasing alloying time and SiC amount [6].

In the study of L. Litynska-Dobrzynska et al, Al-12 wt $\% \mathrm{Zn}-3 \mathrm{wt} \% \mathrm{Mg}-1.5 \mathrm{wt} \% \mathrm{Cu}$ composition ball milled in planetary ball mill by adding $1 \mathrm{wt} \%$ and $2 \mathrm{wt} \% \mathrm{Zr}$ powder. Ball milling was carried out in argon atmosphere with steel balls and ball to powder weight ratio was 10:1. As a result, it leads to a homogeneous mixture of approximately 100 nm grain size nanocrystalline aluminum solid solution after 80 hours of milling. Zr and Cu rich particles of the fcc structure are defined in powders containing Zr [7].

In the study of Indranil Lahiri and Sanjeev Bhargava, $\mathrm{Cu} 10 \mathrm{wt} \% \mathrm{Cr}$ and $\mathrm{Cu} 50 \mathrm{wt} \% \mathrm{Cr}$ alloyed two different compositions in a Pulverisette 5 planetary ball grinder at 200 rpm at different times such as $5 \mathrm{~min}, 1 \mathrm{~h}, 10 \mathrm{~h}, 50 \mathrm{~h}$ and 150 h . While doing this, 5 steel balls having 25 mm diameter with ball to powder ratio $8: 1$ were used. In their study, they made a crystallite size comparison with the data obtained from XRD peaks. According to the results of their studies, crystallite size decreases with the increase of alloying time [8].
M. Rabiee et al. used pure $\mathrm{Fe}(<150 \mu \mathrm{~m})$, pure $\mathrm{Cu}(<150 \mu \mathrm{~m})$ and $\mathrm{SiC}(<12 \mu \mathrm{~m})$ powders as alloying elements. Then $\mathrm{Cu}-20 \mathrm{wt} \% \mathrm{Fe}$ composition was alloyed for 20 hours, by adding $3 \mathrm{wt} \% \mathrm{SiC}$ and 5 $\mathrm{wt} \% \mathrm{SiC}$ alloyed for a further 5 hours in a planetary
ball mill with hardened steel vial and balls under argon atmosphere. With XRD investigation the crystallite size of Cu decreased by increasing milling time and the addition of SiC decreased the crystallite size [9].

In this study, it is aimed to mechanically produce and characterize an alloy similar to aluminum 2024 alloy with a device with spex operating logic.

## 2. Materials and Method

Provide the purity, weight ratios and grain sizes of metal powders and ceramic powder used to produce composite materials with MA treatment in the composition are given in Table 1.

Table 1. Features of Used Powders

| Sample <br> name | Participation Rate of Powders by Weight |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: |
|  | $\mathrm{Zn}(7 \mu \mathrm{~m})$ | $\mathrm{Mg}(45 \mu \mathrm{~m})$ | $\mathrm{Cu}(7 \mu \mathrm{~m})$ | $\mathrm{SiC}(7 \mu \mathrm{~m})$ |  |
| A | $93,2 \%$ | $0,3 \%$ | $1,5 \%$ | $0 \%$ | $5 \%$ |
| B | $90,2 \%$ | $0,3 \%$ | $1,5 \%$ | $3 \%$ | $5 \%$ |
| C | $88,2 \%$ | $0,3 \%$ | $1,5 \%$ | $5 \%$ | $5 \%$ |

These powders were placed in $120 \mathrm{~mm}^{3}$ thermoset plastic bottles with the weight ratios in Table 1, filling $25 \%$ of the volume. 454 steel balls with a diameter of 3 mm were placed in the same bottle, filling $50 \%$ of the bottle volume. The ball / powder weight ratio was applied at a value of 10 . The reason why these numbers are chosen in this way is that they are the values of the best milling and alloying [1].

The bottles were filled with argon gas to prevent oxidation after the powders were placed. The prepared bottles were alloyed on the Retsch MM400 model device at a frequency of 15 Hz for 90 minutes. Before sintering, some amount was taken from the powder mixtures for XRD analysis. The mold parts were coated with Sogerver SP 500 graphite lubricant to prevent sticking of the mold parts and sample before sintering. Sintering process was carried out at $600^{\circ} \mathrm{C}$ for 20 minutes at 200 MPa pressure. After sintering, solid samples were polished to obtain SEM images. The XRD graphic of the samples is shown in Figure 1.


Figure 1. XRD graph of samples
When Figure 1 is examined, the increase in the amount of Cu in the composition based on the highest grade Al peaks caused shorter lengths and widening in the peaks. In Figure 2, SEM images of powder mixtures before alloying are given.


Fig. 1 SEM images of powder mixtures before alloying
When Figure 2 is examined, almost identical images with containing round grains are seen. Other elements gathered around coarse grained Mg can be seen. SiC grains are in a separate powder composition without being integrated with the structure. In Figure 3, SEM images of powder mixtures that are alloyed for 90 minutes are given.


Fig. 3 SEM images of alloyed powder mixtures
When Figure 3 is examined, it is seen that the C sample containing $5 \% \mathrm{Cu}$ is in the pulpy structure. In addition, agglomerates have formed in all sample powders. According to Figure 2, rounded grains turned into flakes. SiC grains are integrated with the structure.

$$
\begin{equation*}
\mathrm{d}=\mathrm{k} \lambda /(\beta \cos \theta) \tag{1}
\end{equation*}
$$

In this equation, d refers to crystal size, k is the shape factor between 0 and $1(\mathrm{k}=0,9), \lambda$ refers to XRD wavelength, $\beta$ refers to full width at half height of the maximum diffraction peak and $\theta$ refers to diffraction angle.

Figure 4 shows the crystallite size of the samples using the data obtained from the XRD graph. Crystallite dimensions were calculated using Scherrer equation (equation 1) [6].


Fig. 4 Crystallite size of samples
When Figure 4 is examined, it is seen that crystallite size decreases with the Cu amount from $0 \%$ to $3 \%$ by weight in the powder composition and the crystallite size does not change after $3 \%$. According to these data taken from Al peaks, it was observed that Cu particles welded to Al particles with $3 \%$ by weight Cu addition, but this welding did not occur $3 \%$ by weight.

## Conclusions

When the XRD graph is analyzed, the highest grade Al peak length of the alloy with $3 \% \mathrm{Cu}$ by weight has decreased compared to the alloy with $0 \% \mathrm{Cu}$ by weight. However, $3 \% \mathrm{Cu}$ by weight, there was no reduction in Al peak lengths.

When the SEM images of the samples were examined, the Al particles, which are the matrix element, were in round shape before alloying, and then became scaly and flat after alloying. It has been observed that SiC particles enter the matrix and take on an integrated structure.

When the crystallite sizes of the samples were calculated, the amount of Cu increased from $0 \%$ to $3 \%$, the crystallite size increased slightly, but the increase in the amount of Cu more weight did not
change the crystallite size.

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